

Materia Optativa de Grado y Postgrado | 2^{do} CUATRIMESTRE 2023

CRISTALOGRAFÍA

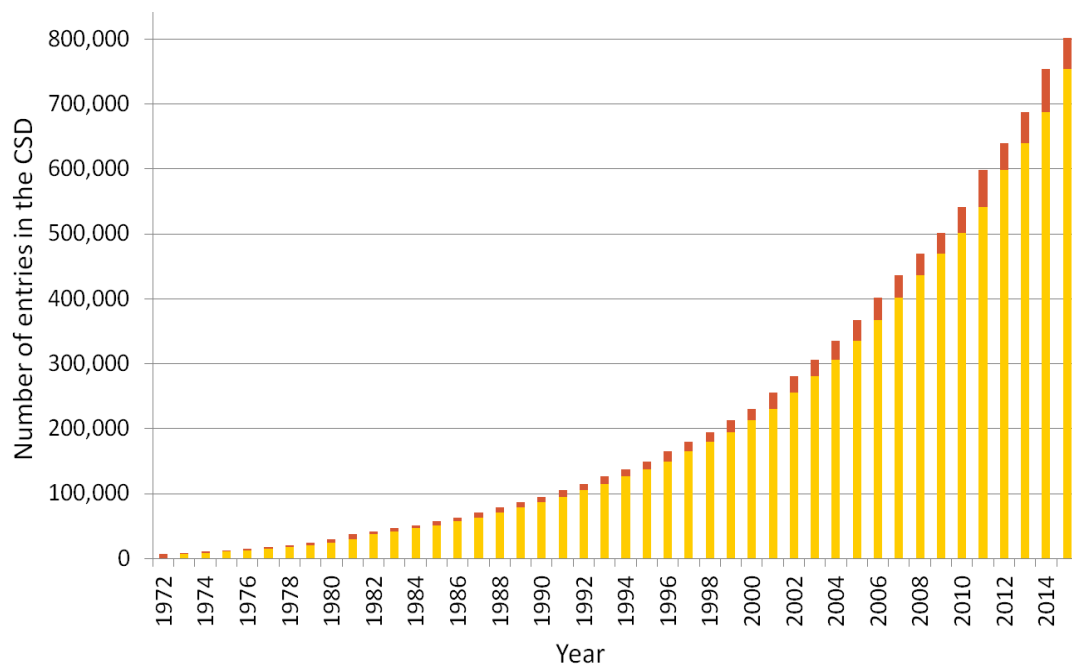
Fundamentos y Aplicaciones

CLASE III

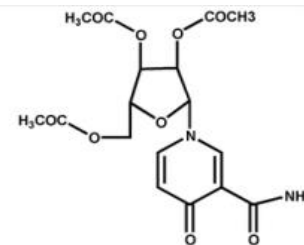
Florencia Di Salvo | Sebastián Suárez

Departamento de Química Inorgánica, Analítica y Química Física, FCEN, UBA

■ Base de Datos

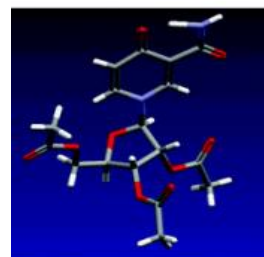


BASYOJ
 4-Oxonicotinamide-1-(1'-beta-D-2',3',5'-tri-O-acetyl-ribofuranoside)
 Source: Rothmannia longiflora
 C17 H20 N2 O9
 G.Bringmann, M.Ochse, K.Wolf, J.Kraus, K.Peters, E.-M.Peters, M.Herderich, L.Ake Assi, F.S.K.Tayman
Phytochemistry (1999) **51**, 271.
 Melting Point: 198-201 deg.C.
 P212121
 a 8.218 b 13.783 c 16.303
 alpha 90.0 beta 90.0 gamma 90.0
 R = 5.6%

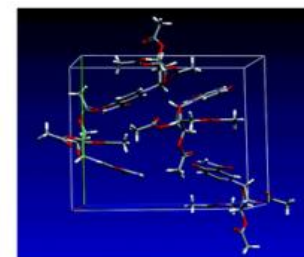


a

b



c



d

Cambridge Structural Data Base (CSD)

La base de datos de Cambridge contiene los datos estructurales cristalográficos derivados de estudios de difracción de rayos X o neutrones en compuestos orgánicos, organometálicos y complejos metálicos. Esta base de datos excluye proteínas, polímeros de alto peso molecular y sales puramente inorgánicas. Cada una de las entradas se identifica con un código de referencia de **8 caracteres (REFCODE)** y contiene información bibliográfica que especifica la referencia de la revista, autores y nombre del compuesto

■ Base de Datos

Understanding Polymorph Stability using Full Interaction Maps

An understanding of the relative stabilities of observed crystal forms may be gained by examining how satisfied the preferred intermolecular interactions are in the lattice.

An understanding of the relative stability of a compound's crystal form can be crucial if it is a candidate for a drug product. There have been occurrences in which a later emergence of a more stable polymorphic form has caused withdrawal of a drug product from the market, costing the manufacturers hundreds of millions of dollars; ritonavir is the classic example of this¹. Use of the Cambridge Crystallographic Data Centre (CCDC) Full Interaction Maps tool can help bolster this understanding and mitigate this risk.

The Full Interaction Maps capability relies upon the

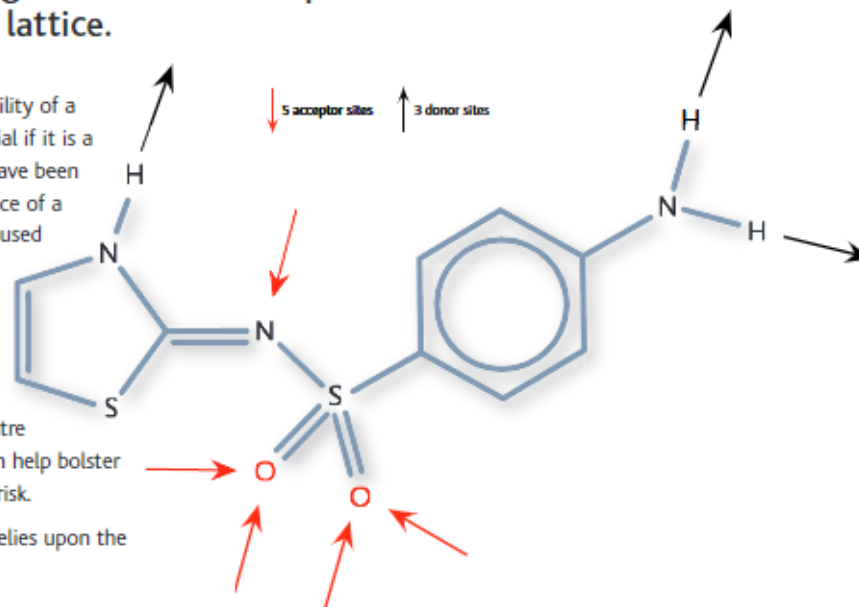


Fig. 1: The 2D chemical structure of sulfathiazole, showing available acceptor and donor interactions

■ Base de Datos



Designing a New Multi-Component API Form Based on a Known Structure

Aim

This use case addresses the topic of how to design new multi-component, crystalline forms of an API purely based on the knowledge of one or more existing forms. The production of new multi-component forms will allow the physico-chemical properties of the solid to be modified (e.g. solubility, crystal habit and stability) without changing the biological efficacy of the API compound. If an isostructural series of API forms can be generated in this way, then tuning of physical properties may even be feasible.

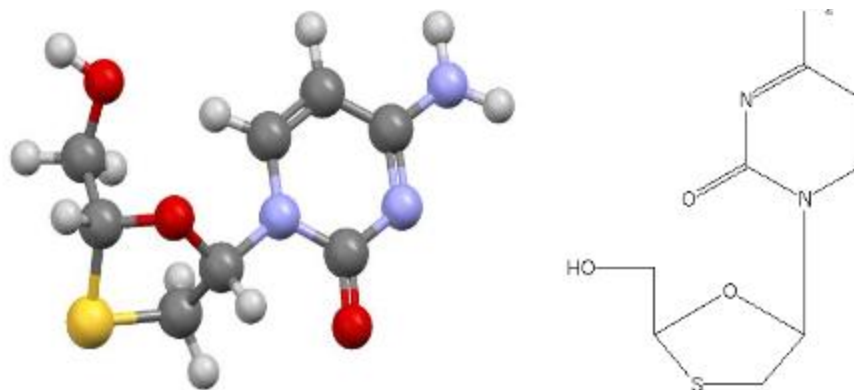
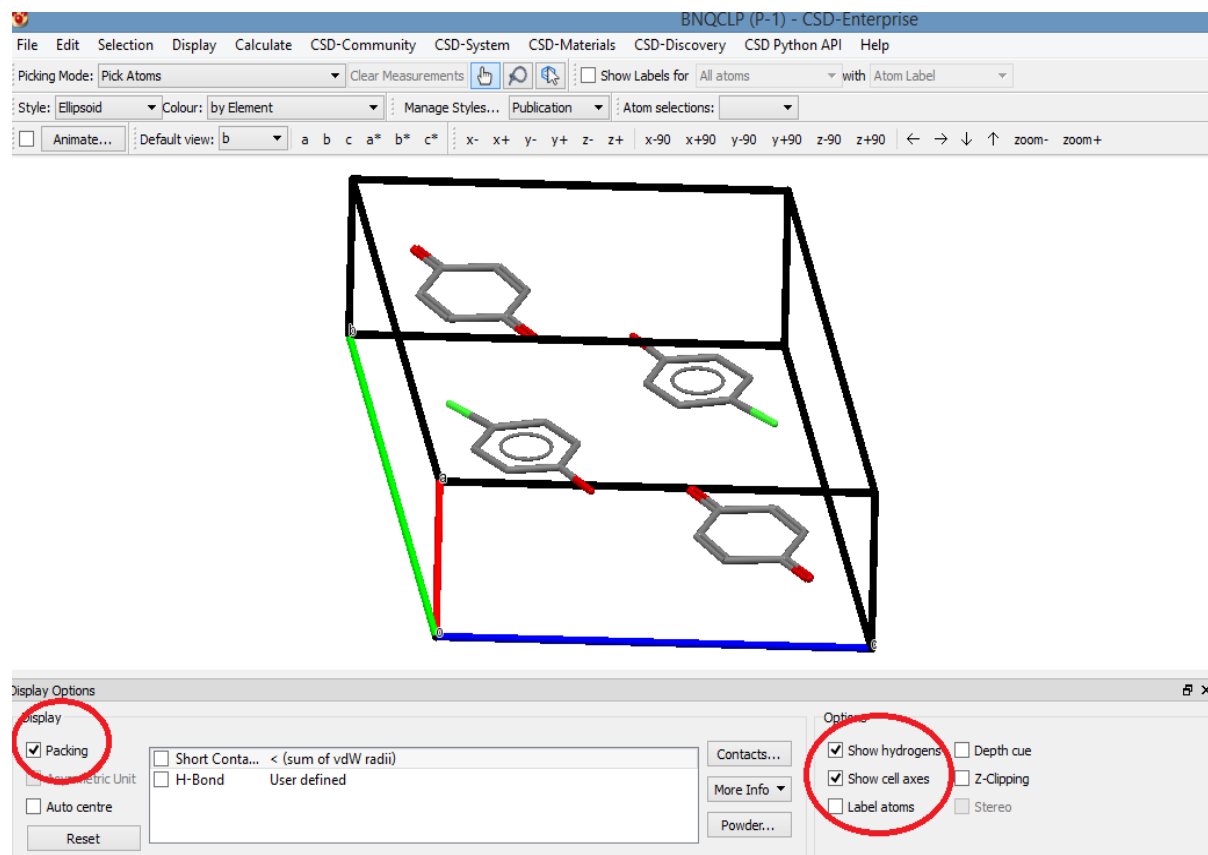
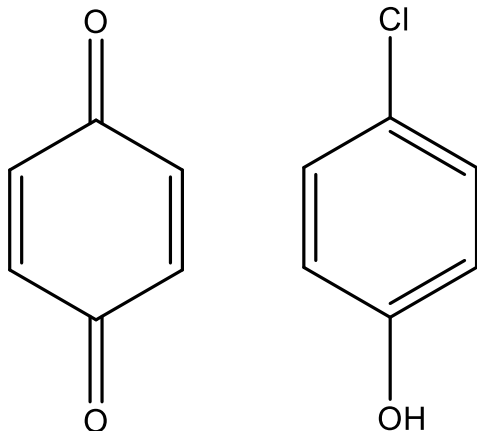


Figure 1 - Lamivudine chemical structure

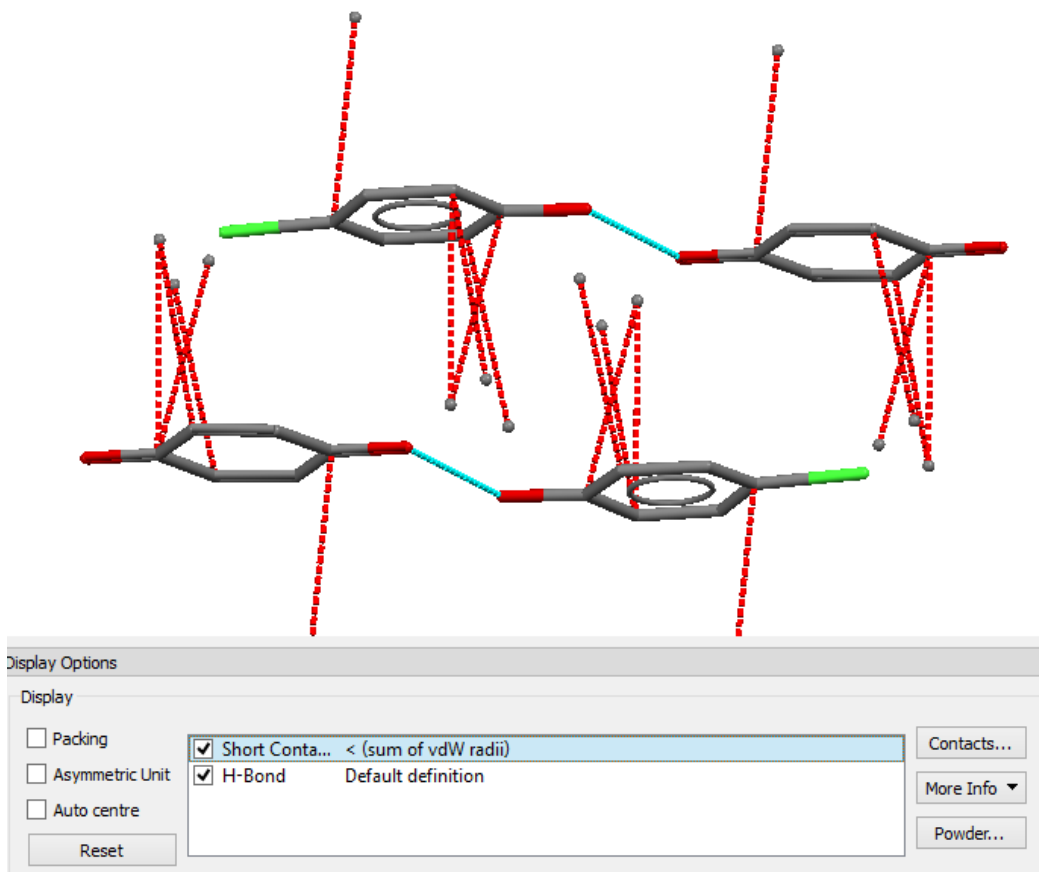
Mercury

Se buscarán interacciones intermoleculares en el co-cristal de *p*-benzoquinona y *p*-clorofenol.



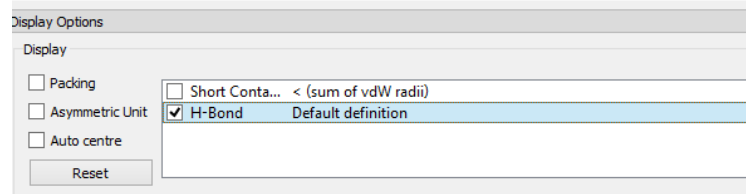
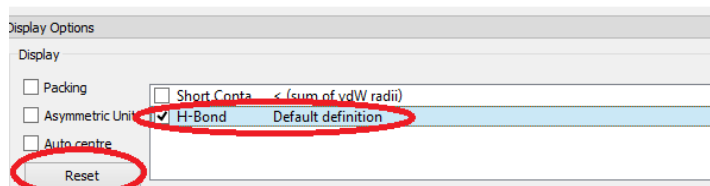
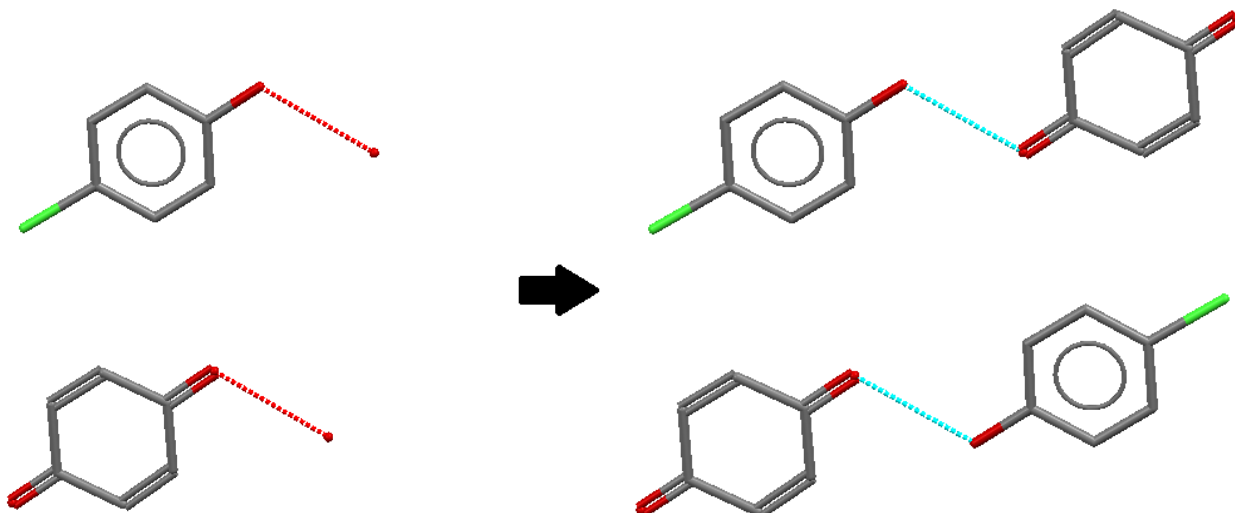
■ Mercury

Se buscaran interacciones intermoleculares en el co-cristal de *p*-benzoquinona y *p*-clorofenol.



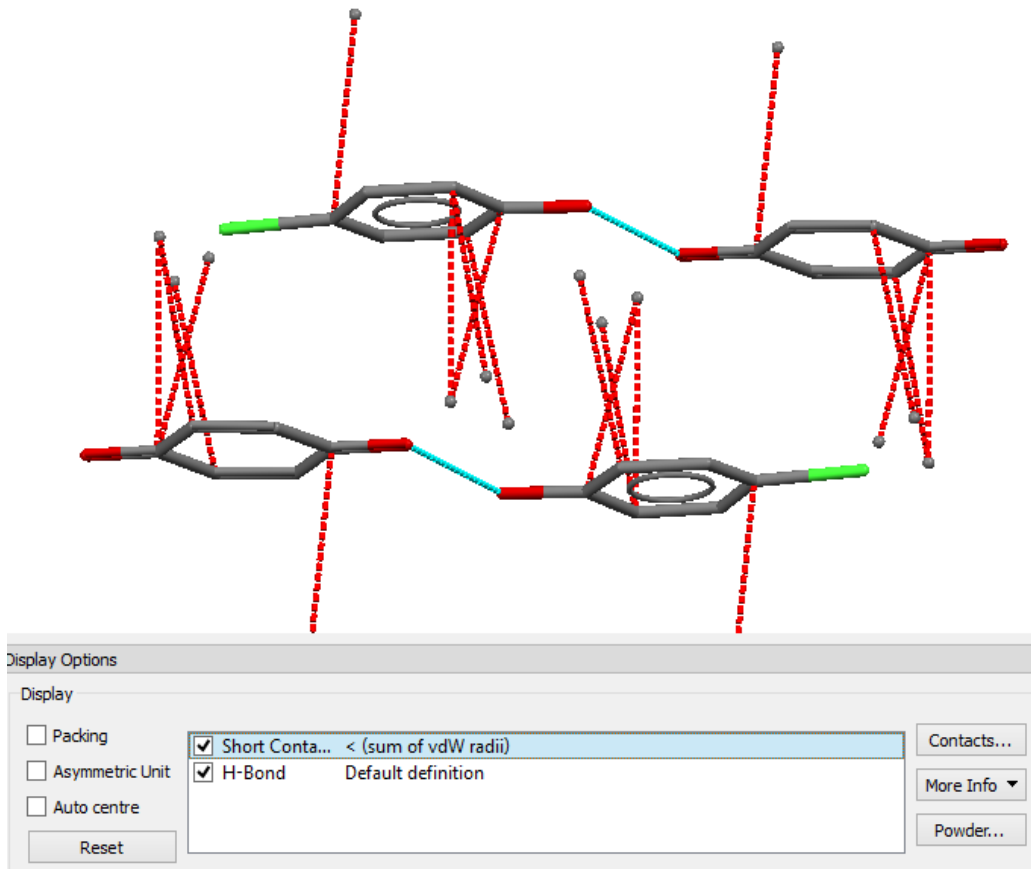
■ Mercury

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■ Mercury

Se buscaran interacciones intermoleculares en el co-cristal de *p*-benzoquinona y *p*-clorofenol.



Mercury

Se buscaran interacciones intermoleculares en el co-cristal de *p*-benzoquinona y *p*-clorofenol.

Current structure: BNQCLP

Right-click on a contact for options. Click on a column heading to sort rows.

Save...

	Number	Atom1	Atom2	Symm. op. 1	Symm. op. 2	Length	Length-VdW	Expanded	Xorth1	Yorth1	Zorth1	Xorth2	Yorth2	Zorth2
1	1	O1	O3	x,y,z	-x,-y,1-z	2.702	-0.338	Yes	-4.3819	1.1889	4.7783	-3.8934	-0.0015	7.1547
2	2	C6	C10	x,y,z	-1+x,y,z	3.385	-0.015	Yes	-3.5493	1.9963	2.7223	-3.1584	-1.2973	2.0455
3	3	C6	C11	x,y,z	-1+x,y,z	3.309	-0.091	Yes	-3.5493	1.9963	2.7223	-4.5156	-0.9679	1.6140
4	4	C7	C10	x,y,z	-1+x,y,z	3.359	-0.041	Yes	-3.2919	1.4443	3.9811	-3.1584	-1.2973	2.0455
5	5	C8	C4	x,y,z	-1+x,y,z	3.308	-0.092	Yes	-2.0272	1.1375	4.3409	-2.9726	-1.8929	3.4119
6	6	C1	C2	x,y,z	-1+x,1+y,z	3.362	-0.038	Yes	-1.2421	1.8541	2.2260	-1.2138	4.8705	3.7092
7	7	C6	C11	x,y,z	-1+x,y,z	3.582	0.132	Yes	-3.5493	1.9963	2.7223	-6.7230	2.0645	1.0621
8	8	C11	C10	x,y,z	-1+x,1+y,z	3.607	0.157	Yes	0.0770	2.0645	1.0621	1.0280	5.4023	2.0455
9	9	C11	O2	x,y,z	1-x,-y,-z	3.379	0.109	Yes	0.0770	2.0645	1.0621	2.2229	1.1152	-1.3687

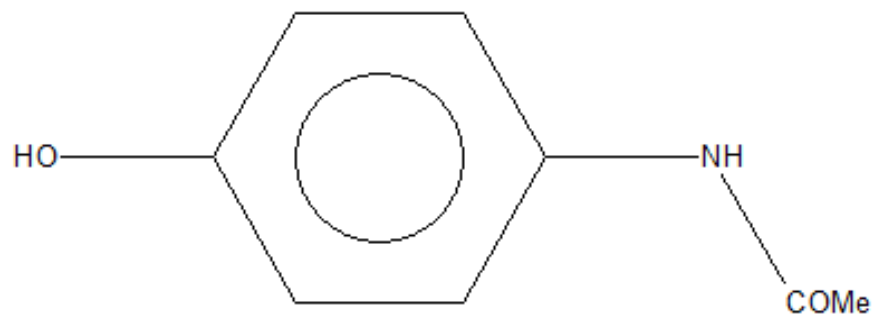
Close

■ Interacciones y Polimorfos

Acetaminofen – Paracetamol

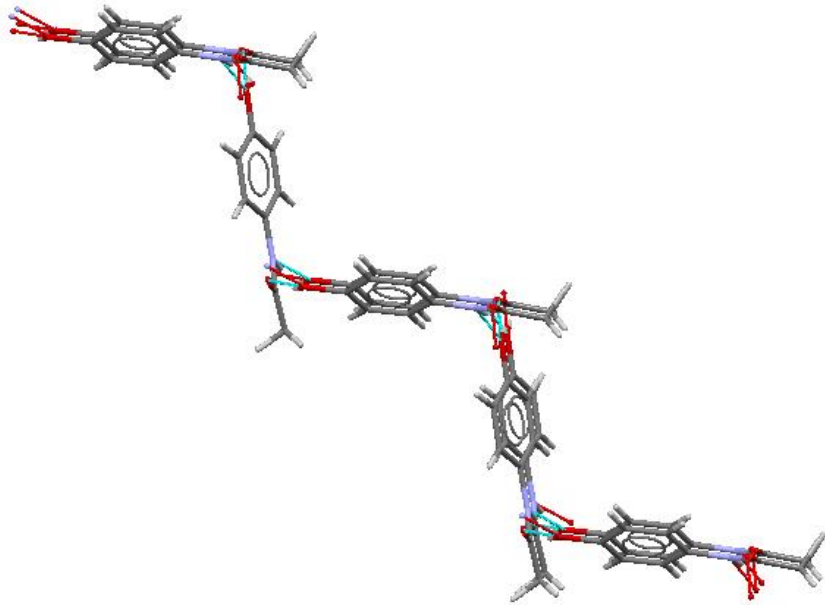
3 Polimorfos principales (Formas I-III)

Más de 25 cocristales e hidratos

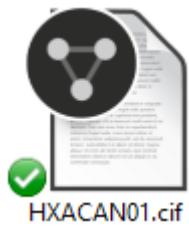
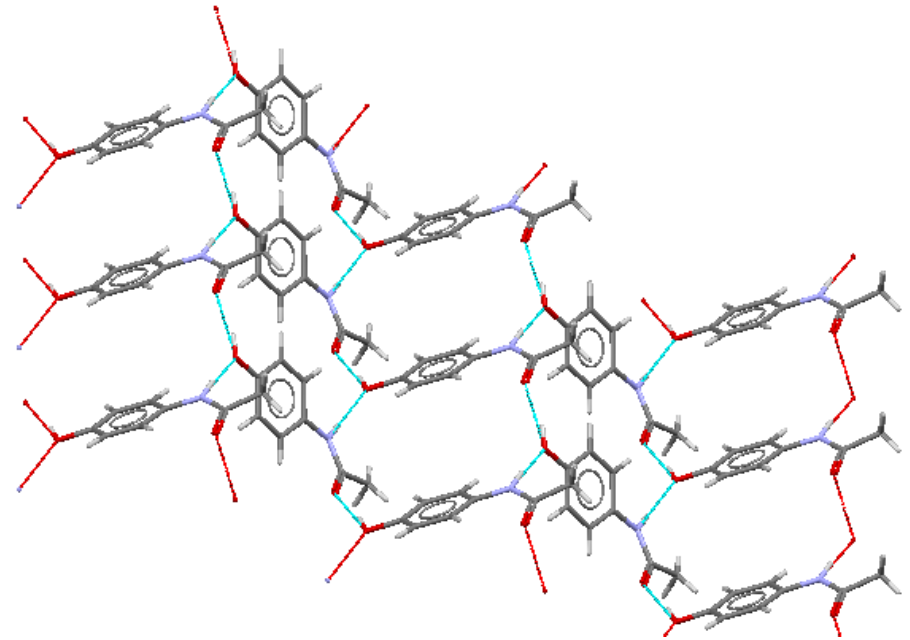


Evaluar las interacciones en los 2 polimorfos, y su packing 3D

■ Interacciones y Polimorfos

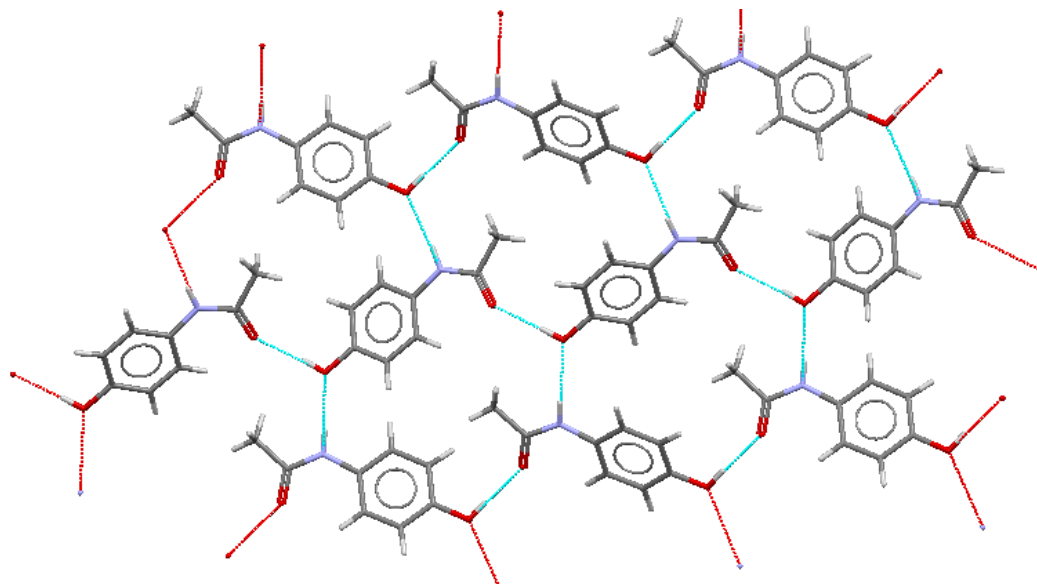


Forma I
monoclínico

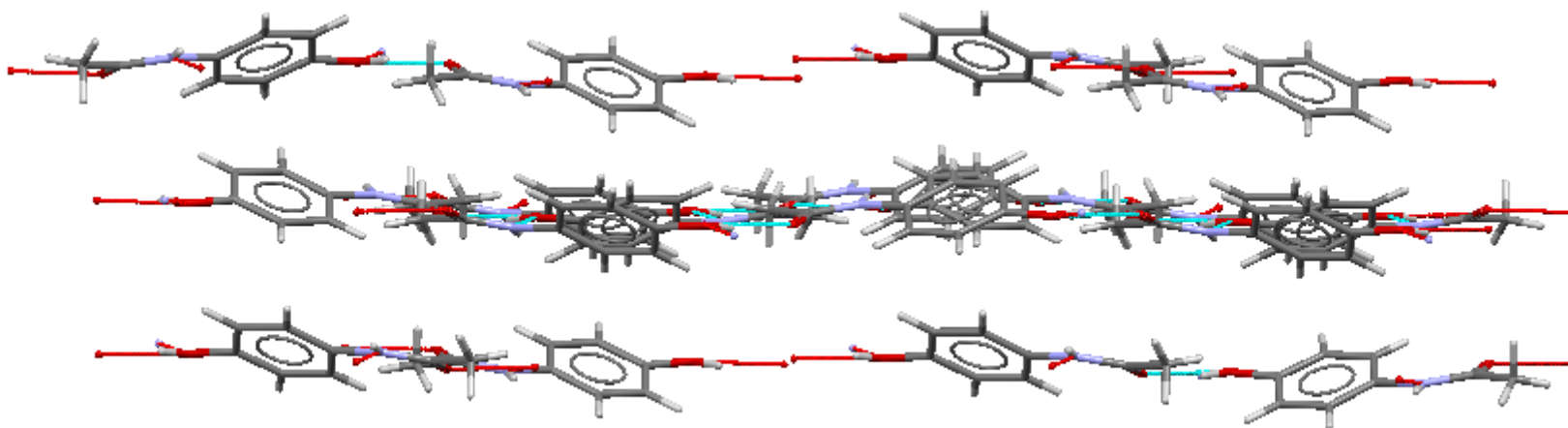


■ Interacciones y Polimorfos

Forma II
ortorrómbico



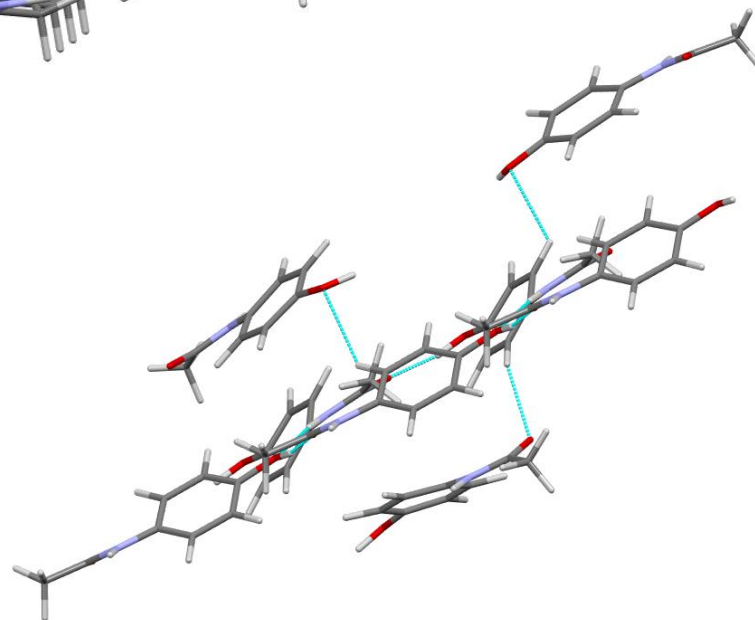
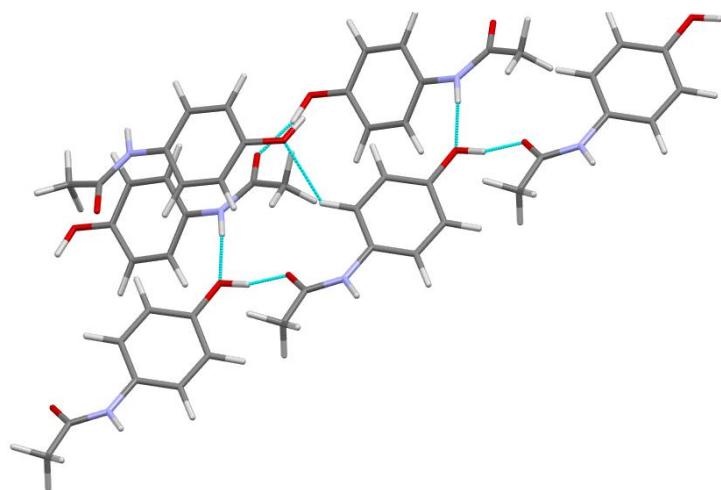
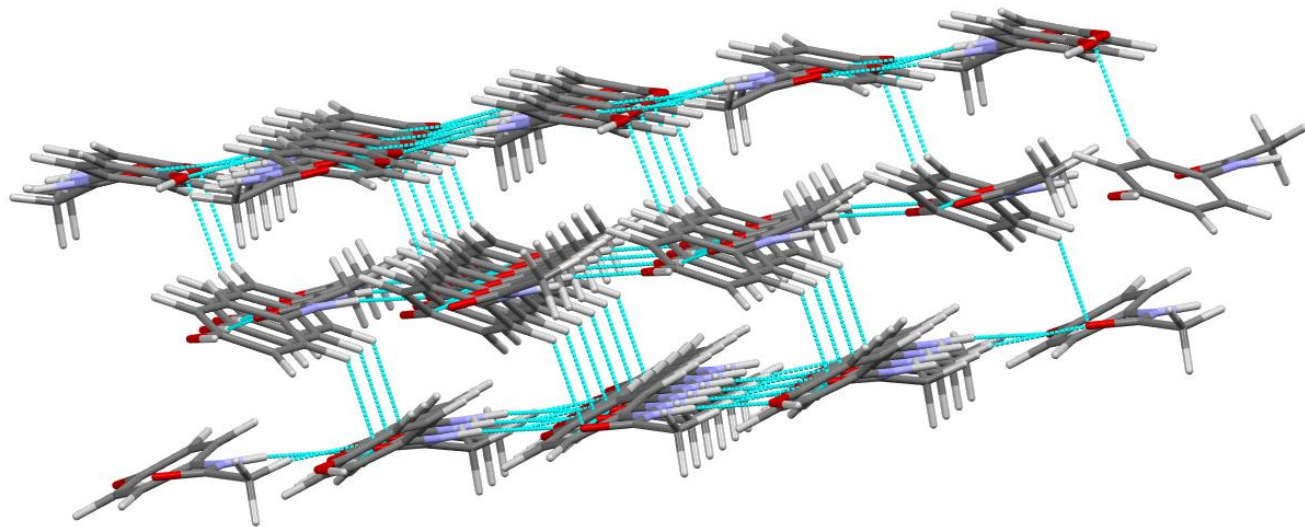
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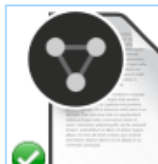
■ Interacciones y Polimorfos

Forma III - metaestable

Ortorrómico (resuelta de datos de DRX de polvo)



HXACAN40.cif



Paracetamol_For
m_III.cif

■ Interacciones y Cocristales

CrystEngComm

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PAPER

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'Masked synthons' in crystal engineering: insulated components in acetaminophen cocrystal hydrates†

Cite this: DOI: 10.1039/c3ce40159f

John R. G. Sander,^a Dejan-Krešimir Bučar,^a Rodger F. Henry,^b Brittany N. Giangigi,^a Geoff G. Z. Zhang^{*c} and Leonard R. MacGillivray^{*a}

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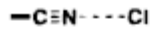
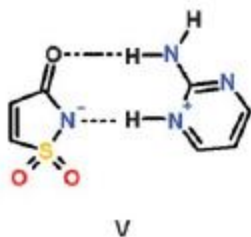
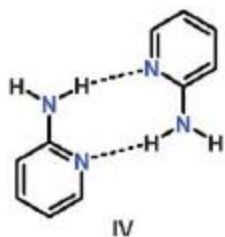
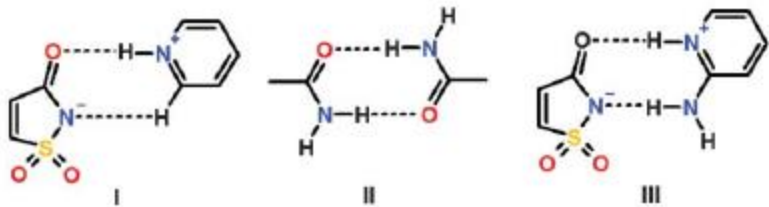
A series of cocrystals of acetaminophen and *trans*-1,2-bis(4-pyridyl)ethylene are reported wherein there is a lack of direct hydrogen-bonding between the components owing to interjected water molecules. A survey of the Cambridge Structural Database demonstrates that interjected water molecules involving cocrystals with similar components are rare. We term the lack of interaction as a 'masked synthon', which we aim to contribute to the development of synthon theory in organic solid-state chemistry.

Sinton

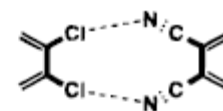


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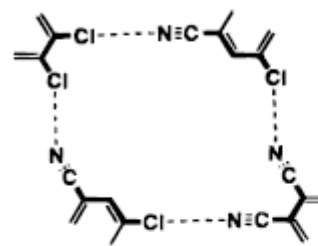
■ Sintón



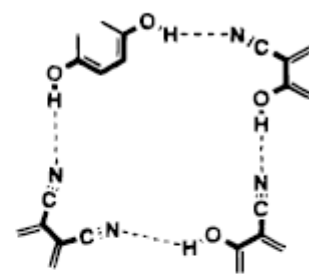
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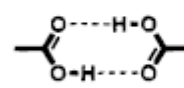
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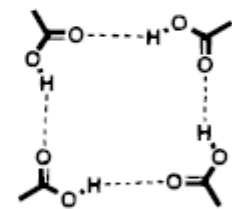
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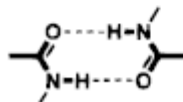
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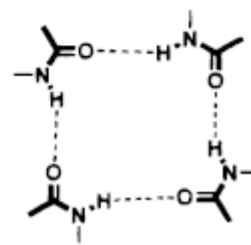
V



VI

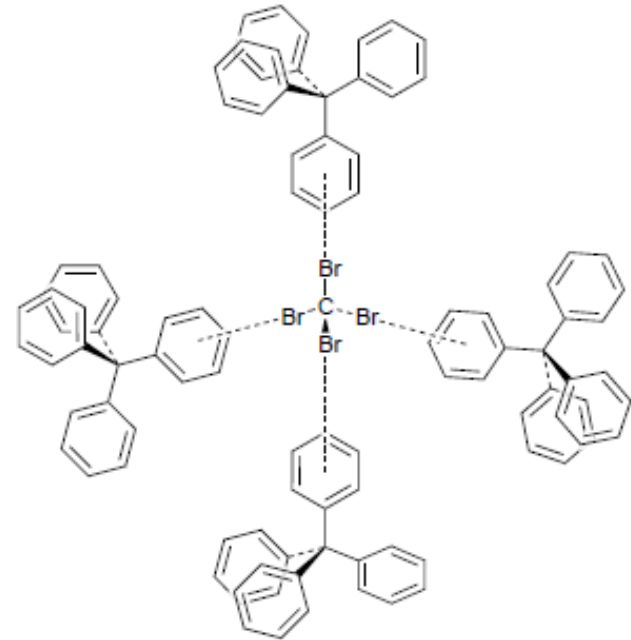
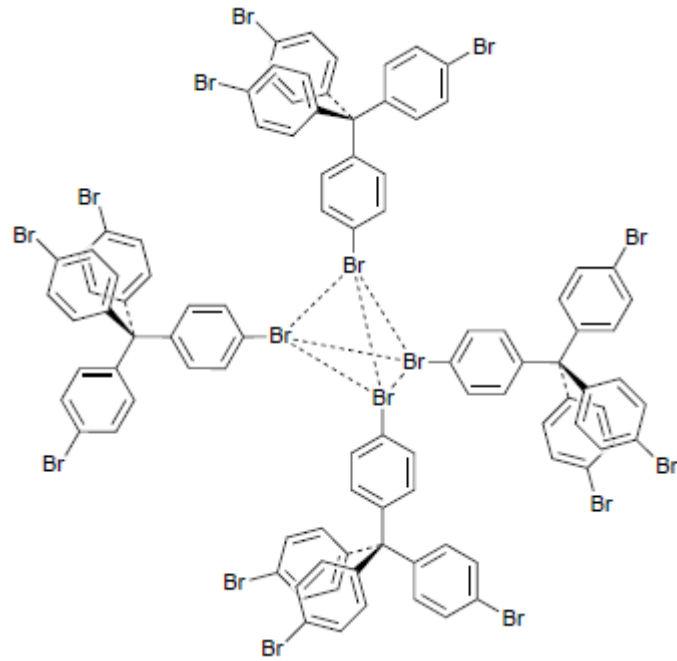


VII



VIII

■ Sinton



■ Interacciones y Cocristales

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'Masked synthons' in crystal engineering: insulated components in acetaminophen cocrystal hydrates†

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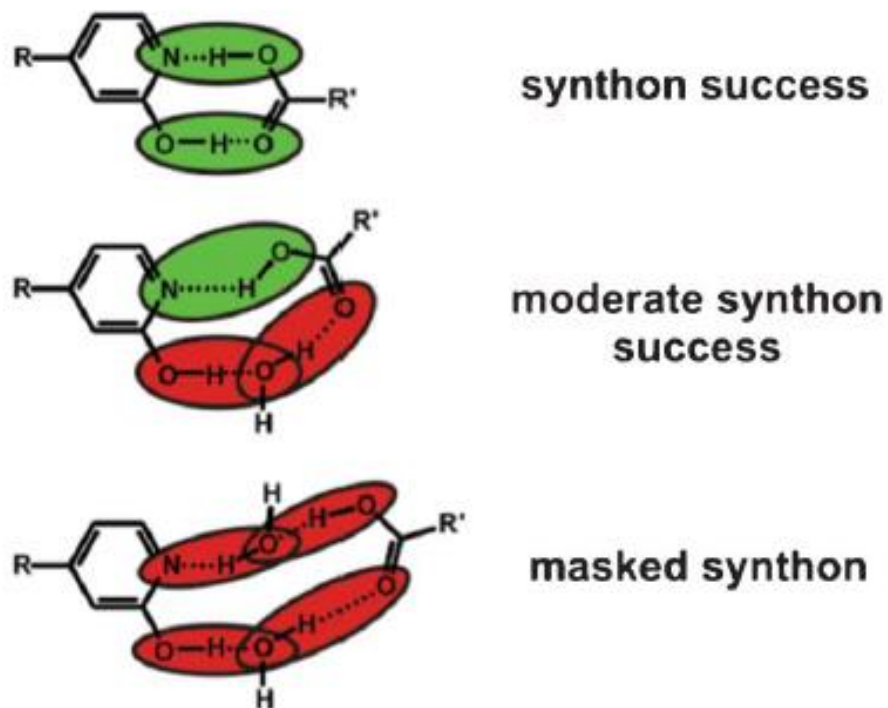
www.rsc.org/crystengcomm

A series of cocrystals of acetaminophen and *trans*-1,2-bis(4-pyridyl)ethylene are reported wherein there is a lack of direct hydrogen-bonding between the components owing to interjected water molecules. A survey of the Cambridge Structural Database demonstrates that interjected water molecules involving cocrystals with similar components are rare. We term the lack of interaction as a 'masked synthon', which we aim to contribute to the development of synthon theory in organic solid-state chemistry.



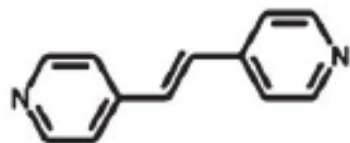
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■ Interacciones y Cocrisales

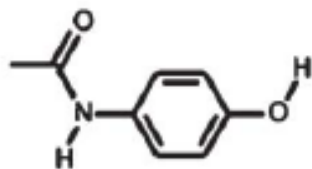


Scheme 1 Comparisons of supramolecular synthons.

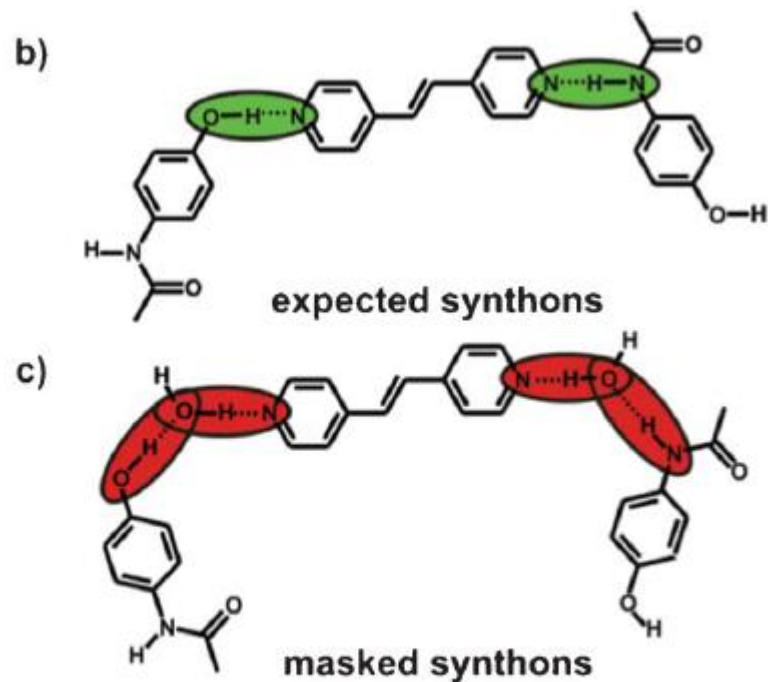
■ Interacciones y Cocrisales



BPE



APAP



Scheme 2 (a) APAP and BPE, (b) expected heterosynthons, and (c) masked synthons formed by interjected water molecules.

■ Interacciones y Cocrisales

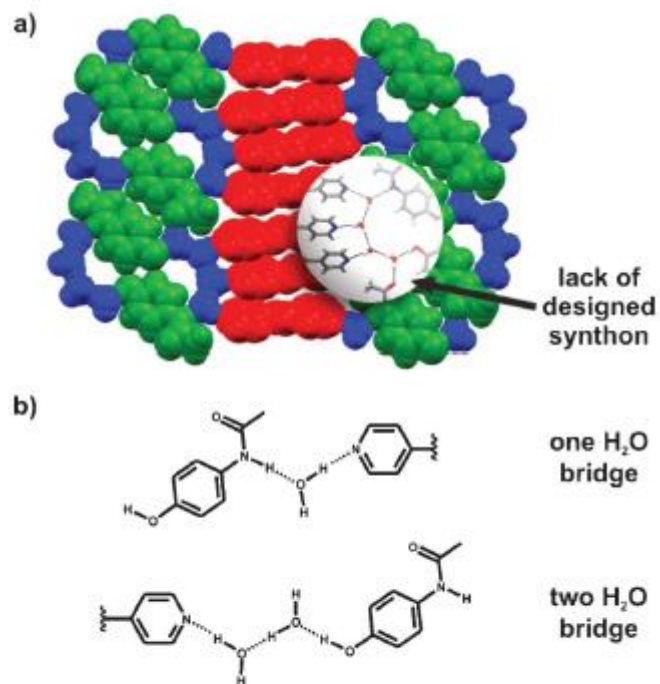


Fig. 1 X-ray structure of **1**: (a) 2D network viewed along a-axis and (b) schematics of water molecules.

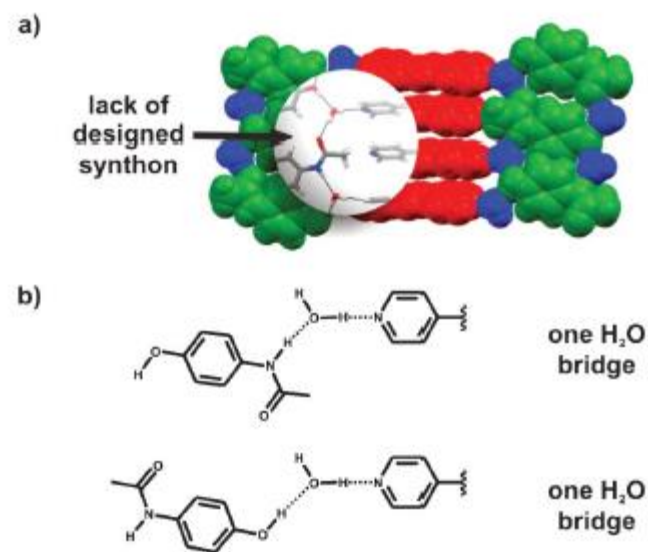


Fig. 2 X-ray structure of **2**: (a) along a-axis to show 2D network (free pyridyl group also present in solid) and (b) water bridges.

■ Interacciones y Cocrisales

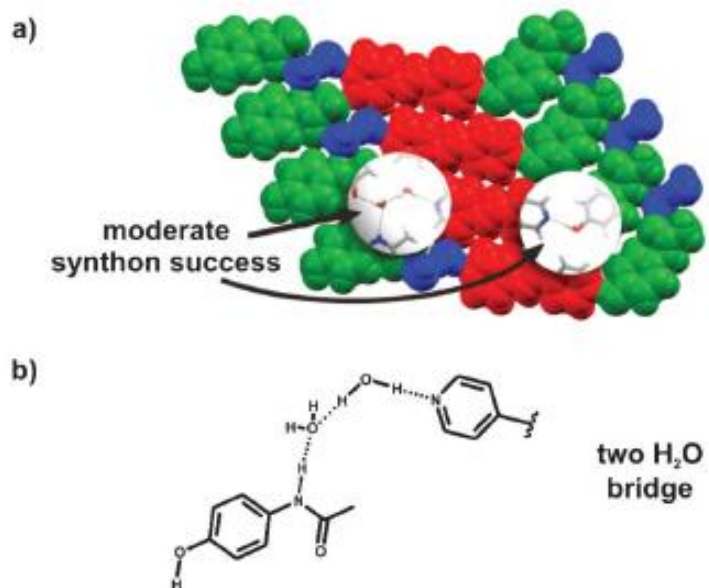


Fig. 3 (a) X-ray structure of **3** along c-axis to reveal 2D sheets composed of wave-like chains and (b) a water dimer that bridges the amide-to-pyridine heterosynthon.



Fig. 4 X-ray structure of **4** showing 1D ladder along c-axis.